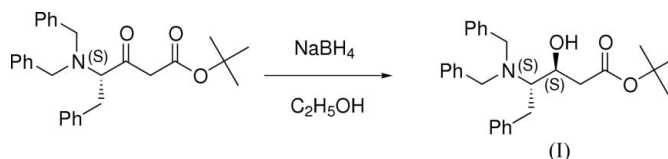
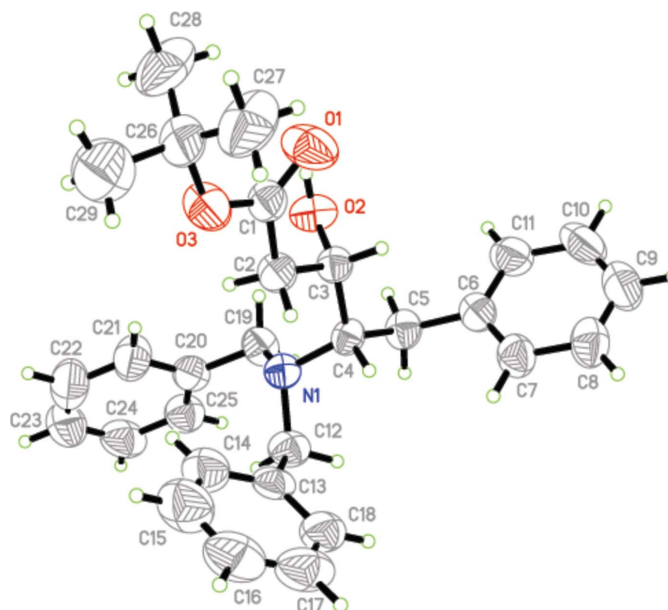
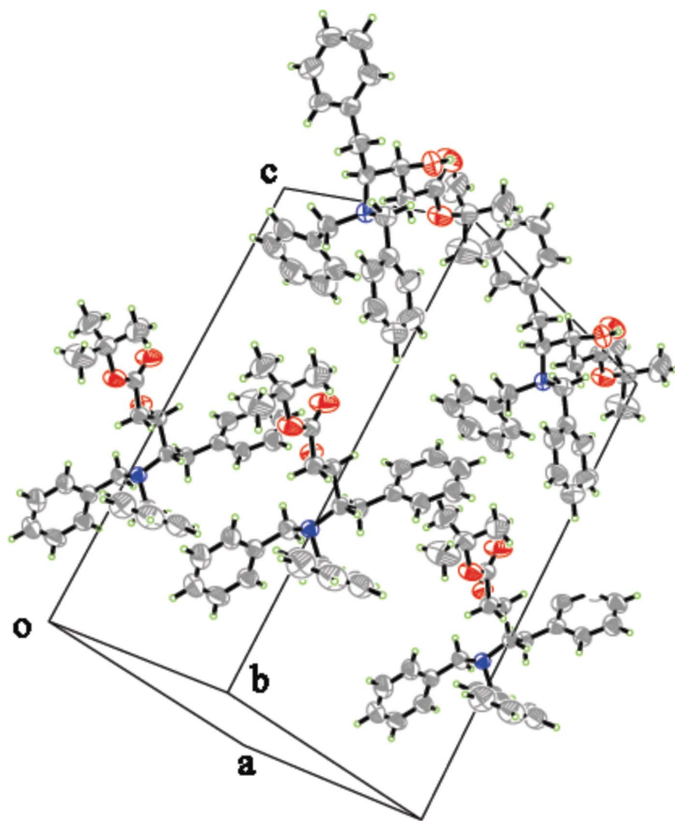


Bin Tan,<sup>a</sup> Jian-Feng Zheng,<sup>a</sup>  
Rong-Bin Huang,<sup>b</sup> Zan-Bin Wei<sup>b</sup>  
and Li-Ren Jin<sup>a\*</sup><sup>a</sup>The Key Laboratory for Chemical Biology of Fujian Province, Department of Chemistry, Xiamen University, Xiamen 361005, People's Republic of China, and <sup>b</sup>State Key Laboratory for Physical Chemistry of Solid Surfaces, Department of Chemistry, Xiamen University, Xiamen 361005, People's Republic of China

Correspondence e-mail: lrjin@xmu.edu.cn

**Key indicators**Single-crystal X-ray study  
 $T = 293\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.007\text{ \AA}$   
 $R$  factor = 0.063  
 $wR$  factor = 0.178  
Data-to-parameter ratio = 10.2For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.**(3*S*,4*S*)-tert-Butyl 4-dibenzylamino-3-hydroxy-5-phenylpentanoate**The title compound,  $\text{C}_{29}\text{H}_{35}\text{NO}_3$ , was obtained as a major product by reduction of (*S*)-*tert*-butyl 4-dibenzylamino-3-oxo-5-phenylpentanoate. The molecular packing in the crystal structure is stabilized by weak intermolecular interactions and van der Waals forces.Received 12 October 2005  
Accepted 21 December 2005  
Online 7 January 2006**Comment**The synthesis of a chiral tridentate ligand, (*3S,4S*)-*tert*-butyl 4-(dibenzylamino)-3-hydroxy-5-phenylpentanoate has been achieved; this is a very important starting material. When we used  $\text{NaBH}_4$  to reduce (*S*)-*tert*-butyl 4-(dibenzylamino)-3-oxo-5-phenylpentanoate (Hoffman *et al.*, 1997), the title compound was formed unexpectedly.**Experimental**(*S*)-*tert*-Butyl 4-(dibenzylamino)-3-oxo-5-phenylpentanoate (443 mg, 1 mmol) was dissolved in ethanol (10 ml). The resulting solution was then cooled to 253 K and treated with  $\text{NaBH}_4$  (114 mg, 3 mmol). The**Figure 1**  
ORTEP 3 (Farrugia, 1997) plot of (I), with displacement ellipsoids drawn at the 50% probability level.



**Figure 2**  
Packing diagram of the title compound (Accelrys, 2001).

reaction was monitored by TLC. After 3 h, the solution was quenched with water (20 ml) and adjusted to pH = 5–6 using 1 M HCl, extracted with diethyl ether, washed with brine (20 ml), dried over MgSO<sub>4</sub> and finally concentrated to obtain the crude product (424 mg). Suitable crystals were obtained by allowing the crude product solution to stand for three days.

#### Crystal data

C<sub>29</sub>H<sub>35</sub>NO<sub>3</sub>  
M<sub>r</sub> = 445.58  
Orthorhombic, *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>  
a = 10.351 (3) Å  
b = 12.615 (3) Å  
c = 19.640 (5) Å  
V = 2564.5 (11) Å<sup>3</sup>  
Z = 4  
D<sub>x</sub> = 1.154 Mg m<sup>-3</sup>

Mo Kα radiation  
Cell parameters from 4261 reflections  
θ = 2.6–22.0°  
μ = 0.07 mm<sup>-1</sup>  
T = 293 (2) K  
Chunk, colorless  
0.56 × 0.28 × 0.18 mm

#### Data collection

Bruker APEX area-detector diffractometer  
φ and ω scans  
Absorption correction: multi-scan (SADABS; Bruker, 2001)  
T<sub>min</sub> = 0.960, T<sub>max</sub> = 0.987  
14713 measured reflections

2994 independent reflections  
2600 reflections with I > 2σ(I)  
R<sub>int</sub> = 0.030  
θ<sub>max</sub> = 26.5°  
h = -12 → 12  
k = -15 → 13  
l = -24 → 24

#### Refinement

Refinement on F<sup>2</sup>  
R[F<sup>2</sup> > 2σ(F<sup>2</sup>)] = 0.063  
wR(F<sup>2</sup>) = 0.178  
S = 1.12  
2994 reflections  
293 parameters  
H-atom parameters constrained

w = 1/[σ<sup>2</sup>(F<sub>o</sub><sup>2</sup>) + (0.0835P)<sup>2</sup> + 0.6494P]  
where P = (F<sub>o</sub><sup>2</sup> + 2F<sub>c</sub><sup>2</sup>)/3  
(Δ/σ)<sub>max</sub> < 0.001  
Δρ<sub>max</sub> = 0.57 e Å<sup>-3</sup>  
Δρ<sub>min</sub> = -0.33 e Å<sup>-3</sup>

C29 was not refined anisotropically due to unresolved disorder. H atoms were positioned geometrically (C–H = 0.93, 0.93, 0.96 and 0.97 Å for phenyl, methine, methylene and methyl H atoms, respectively, O–H = 0.82 Å). U<sub>iso</sub>(H) values were set equal to xU<sub>eq</sub> of the carrier atom, where x = 1.5 for methyl and x = 1.2 for all other H atoms. In the absence of significant anomalous scattering effects, Friedel pairs were merged. The absolute configuration was assigned on the basis of the known configuration of (*S*)-*tert*-butyl 4-(dibenzylamino)-3-oxo-5-phenylpentanoate.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997) and ViewerPro (Accelrys, 2001); software used to prepare material for publication: SHELXL97.

The authors thank the Fujian Science Foundation and Xiamen Science Foundation for financial support. We also thank the Key Laboratory for Physical Chemistry of Solid Surfaces for providing the X-ray diffraction facilities.

#### References

- Accelrys (2001). *ViewerPro*. Version 4.2. Accelrys Inc., Burlington, Massachusetts, USA.  
Bruker (2001). *S*AINT (Version 6.22), *S*MART (Version 5.625) and *S*ADABS (Version 2.03). Bruker AXS Inc., Madison, Wisconsin, USA.  
Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.  
Hoffman, R. V., Tao, J. H. (1997). *J. Org. Chem.* **62**, 2292–2297.  
Sheldrick, G. M. (1997). *S*HELXS97 and *S*HELXL97. University of Göttingen, Germany.